Some Physical Constants

	100	3.00 × 10° m/s
Spood of light	6	6.67 × 10 ⁻¹³ N - m ³ /kg
Gesvitational constant	100	8.99 × 10° N · m ² /C ²
Coulomb's constant		6.63 × 10 ⁻³⁴ J · s
Planck's cometant	100	1.58 × 10 · m J/K
Boltzmann's constant	100	1.60 × 10 ⁻¹⁷ C
Elementary charge	100	9.11 × 10 ⁻²⁰ kg
Electron muse	- 33	1.67 = 10 ⁻²⁷ kg
Photosa rease	100	1.68 × 10 ⁻²⁷ kg
Neutron mans	1000	

Commonly Used Physical Data

Gravitational field strongth (9.80 J - kg - m 9.80 m/s
Mass of the earth My	5.98 × 10 ²⁴ kg
Radius of the earth R.	6380 km (equatorial)
Mass of the sun M.	1.99 × 10 ³⁰ kg
Radius of the sun R.	696,000 km
Distance to moon	3.84 × 10° m
	1.50 × 10 ¹¹ m
Distance to sun	5000.0 kg/m ³ = 1 g/cm ³
Density of water	1.2 kg/m ³
Density of air	0.K = -273.15°C = -459.67°F
Absolute zero	U. P. In the Late of the Control of
Precing point of water?	273.15 K = 0°C = 32°F
Room temperature	295 K = 22°C = 72°F
Soiling point of water	373 K = 100°C = 212°F
Normal atmospheric pressure	100.3 kPa

Standard Metric Prefixes

For powers of 10

Power	Prefix	Symbol
10	10000	E
1025	preta:	P
10 ¹⁸ 10 ¹³	Berrie	T
107	polygon	G
1/0"		3-6
107	tinega killo	16.2
10-2	coentii	k.
10	1000000	
10 -	micro	(DE)
100	That the	100
101-10	price	P.
100 175	ternto	100
207-10	distant	46

Useful Conversion Factors

1 meter = 1 m = 100 cm = 39.4 in. = 3.28 ft
1 mile = 1 mi = 1609 m = 1.609 km = 5280 ft
1 inch = 1 in. = 2.54 cm
1 light-year = 1 ly = 9.46 Pm = 0.946 × 1014 m
1 hour = 1 h = 60 min = 3600 s
1 day = 1 d = 24 h = 86.4 ks = 86,400 s
The state of the s
1 year = 1 y = 365.25 d = 31.6 Ms = 3.16 × 10"
1)=1kg·m ¹ /s ²
1 N - m = 0.239 cal
1 cal = 4.1861
1W-11/s
1 kW - h = 3.6 MI
1 K (temperature difference) = 1°C = 1.8°F
The second secon
- (1K)
$T = \left(\frac{1 \text{ K}}{1 \text{ CC}}\right) (T_{(C)} + 273.15 \text{ C})$

1 m/s = 2.24 mi/h
1 mi/h = 1.61 km/h
1 ft³ = 0.02832 m³
1 gallon = 1 gal = 3.79 × 10⁻³ m³ = 3.8 kg H₂O
1 N = 1 kg : m/s² = 1 J/m = 0.225 ft³
1 ft = 4.45 N
Weight of 1-kg object near earth = 9.8 N = 2.2 ft³
1 cal = energy to raise temperature of 1 g of H₂O by 1 K
1 food calorie = 1 Cal = 1 kcal = 1000 cal
1 horsepower = 1 bp = 746 W
1 Pa = 1 N/m²

$$T_{|Cl} = \left(\frac{5 \cdot C}{9 \cdot F}\right) (T_{|Cl} = 32 \cdot F)$$

 $T = \left(\frac{5 \cdot K}{9 \cdot F}\right) (T_{|Cl} = 459.67 \cdot F)$

 $1 \text{ pey} = 360^{\circ} = 2 \text{ medians} = 6.28 \text{ rad}$

1 rad = 57.3° = 0.1592 nev

Commenced strongstonic processes

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Michael Ware, Justin Peatross

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