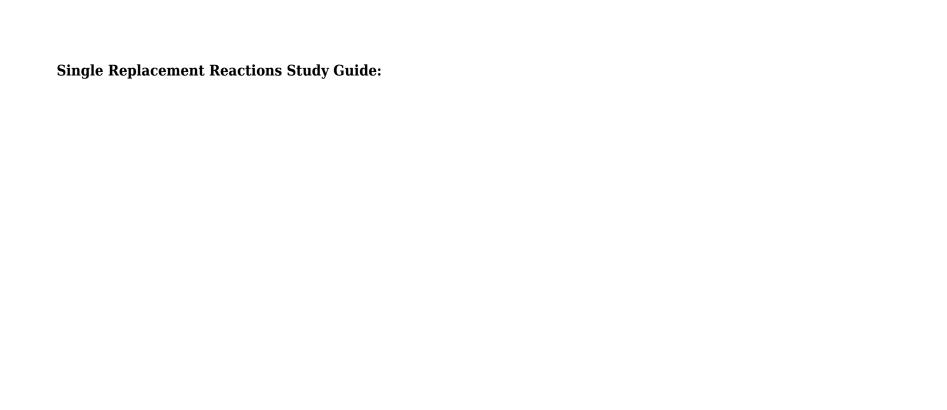
General Chemistry Mr. MacGillivray Worksheet: Predicting Single Replacement Reactions

For each of the following reactions,

- PREDICT whether or not the reaction will occur. Use the activity series to determine this. If the reaction will not occur, write "NO REACTION."
- If the reaction occurs, WRITE THE CORRECT FORMULAS for the reactants and products.
 - Remember: the following elements are diatomic: H₂, N₂, O₂,
 F₂, Cl₂, Br₂, I₂.
 - Use the chart of polyatomic ions and transition metals if necessary.
- BALANCE the equation.
- #1 and #5 are done already as examples.
- Ni + CuCl₂ → Cu + NiCl₂
- Zn + Pb(NO₃)₂ →
- 3. Cl₂ + KI →
- 4. Cu + FeSO₄ →
- 5. Ba + HOH → Ba(OH)₂ + H₂
- calcium + sodium nitrate →
- 7. zinc + lead (II) nitrate →
- 8. aluminum + mercury (II) acetate →
- 9. Al + NiSO₄ →
- 10. sodium + water →

Single Replacement Reactions Study Guide

Chao Zhang



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